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Effect of Iron Toxicity on Rice Growth in Sulfato-ferruginous Lowland of South Senegal

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Abstract: Rice production in southern Senegal is mainly rain-fed and faces numerous constraints, including iron toxicity. The objective of this study was to determine the effect of iron toxicity on rice yield in Southern Senegal. The study was carried out in farmers' fields in the area of Sindone where two sites were selected (Sites A and B). In each site, six farmers' rice fields covering 2000 to 2500 m² were selected: three fields with apparent iron oxide (rust) located in lowland and three fields without iron oxide located in highland. Within each of these fields, three plots with dimensions of 300 m² were delimited and used for the experiment for a total of 18 plots (6*3). Soil cores were taken from all plots in the 0 – 10 cm horizon at 0, 30, 60 and 90 days after rice transplanting (DAT) to measure soil pH. The intensity of iron toxicity was evaluated on rice plants at different dates and rice yield was determined at harvest at 90 DAT. At 0 and 90 DAT, pH was more acidic compared to 30 and 60 DAT for all plots. Besides, in plots affected by iron oxide, pH at 0 and 90 DAT was statistically more acidic ($p < 0.01$) than pH in non-affected plots. In those plots, despite normal growth and tillering, it was only at 90 DAT that brown spots appeared at the tip of the oldest leaves. In plots affected by iron oxide, at 30 DAT, symptoms of iron toxicity appeared and intensified at 60 and 90 DAT; growth and tillering were reduced with many leaves becoming discolored at both sites. Rice yield decreased by at least 43% in plots affected by iron oxide for sites A and B. Amendments that will reduce soil acidity would be recommended, this will also improve rice nutrients and increase rice yield.

Keywords: Paddy Fields, Acidity, Rust, Iron Toxicity, Rice Yield

1. Introduction

In Senegal, and more particularly in Casamance, rice (*Oryza sativa* L. and *Oryza glaberrima* Stend) is grown mainly in sulfato-ferruginous lowlands derived from fluviomarine sediments. Iron is one of the most abundant microelement in tropical ferritic and ferruginous soils [1]; [2]. Iron in higher elevation are leached toward the lowlands [3]. In highlands and slopes, iron is generally in the ferric form [Fe³⁺]. This form is insoluble and therefore

unavailable to plants. In the lowlands, soil which is engorged with water is poor in oxygen and tends to be acidic [4]. In those lowlands, ferric iron is then reduced to ferrous iron [Fe²⁺] [5]. The presence of ferrous iron or iron oxide (rust) is visible through rusty soil and oily stain in water. In the soil, the excess of iron oxide is shown by red-orange spots. The latter is soluble in water and therefore available for plants that need it in very small quantities. This form of iron, easily absorbable, results in toxicity of plants [6]. The flooding of rice fields thus favors the reduction of iron, which becomes toxic to rice [7]. Iron

toxicity, also known as *yellowing* or *bronzing*, can be defined as a nutritional disorder associated with high concentrations of iron in the soil solution [8].

Symptoms of iron toxicity are small reddish to brown spots on the leaves. The *bronzing* is used as an indicator of iron toxicity [9]. These spots extend, merge and give a reddish color to the leaves [7, 10, 11]. Manifestations of iron toxicity may be associated with deficiency of phosphorus and zinc; and a decreased in soil chemical fertility [12].

Iron toxicity alters root structure, delays phenological development and causes sterility of panicles [13, 14]. The symptoms are often associated with limited rice growth (height and tillering), and lower yield [15, 16]. Rice yield can thus fall from 10% to 100% depending on the concentration of iron and the tolerance of the cultivar [16, 17]. However, rice with *bronzing* does not always result in a decrease of yield [18, 19]. With the rainy season, the flooding of paddy fields causes a rapid and significant increase in ferrous iron, lowering the pH and causing a release of more iron [7, 20]. Over time, with waterlogging, the environment become more acidic with the decomposition of organic matter [7]. The effects of iron toxicity is more pronounced at the end of the rainy season. Many strategies to improve rice yield have been adopted including the use of selected tolerant rice varieties [21, 22], associated with better soil management practices [23]. However, the yield remain low and more land are being affected. In South Senegal more paddy fields are being affected by the presence of iron; however, the impact of this iron toxicity on rice production remains unknown. Determining the impact on rice yield in relation to soil acidity would be essential to carry out better management strategies. This study was carried out to determine the effects of toxicity on rice yield in the South of Senegal.

2. Materials and Methods

2.1. Presentation of the Study Area

The study area was located in southern Senegal in the village of Sindone, located on the left bank of the Casamance River. The climate is characterized by two seasons: a rainy season of 3 months (July-September) and a dry season of 9 months. The average rainfall is around 1300 mm. Annual average temperatures are 32°C during the day and 24°C at night. The landscape consists of trees and shrubs on a ferrosol type with a clay-silty texture.

2.2. Experimental Design and Data Collection

The local rice variety Kebo was used. The rice was first sown in a nursery for 21 days. Then, the rice was transplanted on the lowlands and the highlands in August 2016.



Figure 1. Water in paddy fields affected by iron oxide

The experiment was set up in two sites A and B in farmers' fields located in Sindone. In each site, six (6) rice fields with a size of 2000 to 2500 m² were chosen for the experiment. A split-plot design was set up with the presence or absence of iron oxide in the two sites. Three of these fields were located in lowland (PfeOx1, PfeOx2, PfeOx3) with the presence of iron oxide (red water with oily or rust spots (Figure 1)). The other three fields, located on the highland were considered as plots not affected by iron oxide (P1, P2, and P3). Three plots of 300 m² were delimited within each field for data sampling for a total of 18 plots (6*3). The pH of the water and the soil was measured using a pH-meter at the day of rice transplanting corresponding to 0 days after transplanting (DAT), at 30 DAT, at 60 DAT, and 90 DAT. Samples of soil and rice plants were collected from each plot at 0, 30, 60 and 90 DAT.

The effect of iron toxicity on the leaves was evaluated by direct observation of the leaves. A grade was assigned according to the standard scale established by the International Rice Research Institute (IRRI) [24] (Table 1). This scale, based on leaf symptoms, varies from 1 to 9 depending on the color.

Table 1. A standard scale of the Iron toxicity Intensity assessment (IRRI) [24].

Index	Description
0	Growth and tillering nearly normal
1	Growth and tillering nearly normal; reddish-brown spots or orange discoloration on tips of older leaves
3	Growth and tillering nearly normal, older leaves reddish-brown, purple, or orange-yellow
5	Growth and tillering retarded, many leaves discolored
7	Growth and tillering ceases, most leaves discolored or dead
9	Almost all plants dead or dying

At 90 DAT, rice yield was estimated per hectare. The formula of [25] was used.

$$\text{Yield} = \text{NP} \cdot \text{ha}^{-1} \times \text{NT} \cdot \text{P}^{-1} \times \text{Npa} \cdot \text{T}^{-1} \times \text{NG} \cdot \text{Pa}^{-1} \times \text{PG}$$

Where NP.ha⁻¹ = number of plants / hectare = NP.m⁻² x 10,000

NT.P⁻¹ = number of tillers / plant;

Npa.T⁻¹ = number of panicles / tillers;

NG.Pa⁻¹ = number of grains / panicles;

PG = mass of a grain.

2.3. Statistical Analyses

The data was analyzed using the XLSTAT 2014 version 5.03 software. ANOVA was used to analyze the variances of means. The difference in means was subsequently established by Fisher test at the 5% level.

3. Results

3.1. Soil pH

Soil pH in plots with iron oxide was significantly lower ($P < 0.001$) compared to plots without iron oxide at 0 and 90 DAT regardless of the site (Figures 2 & 3). At 0 DAT, plots of the Site A, affected by iron oxide, had an average pH of 4.29 compared to plots not affected by iron oxide where the pH was 5.34. In site B, the mean pH was 4.51 in plots with iron oxide against 5.37 for plots without iron oxide.

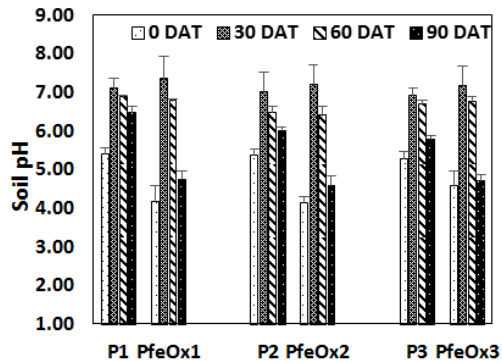


Figure 2. pH data from plots affected and non-affected by iron oxide for site A at 0, 30, 60 and 90 days after rice transplanting (DAT).

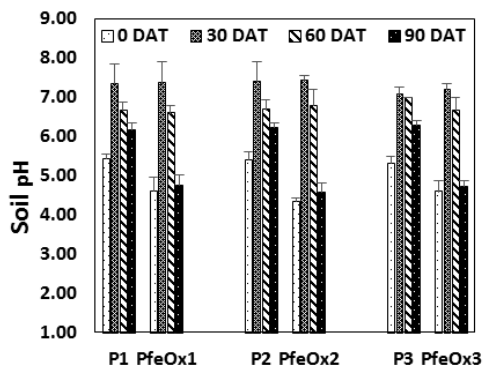


Figure 3. pH data from plots affected and non-affected by iron oxide for site B at 0, 30, 60 and 90 days after rice transplanting (DAT).

There was no significant difference for the pH at 30 and 60 DAT between plots with iron oxide and without iron oxide regardless of the site. The pH at 30 and 60 DAT were significantly higher than mean pH at 90 DAT, regardless of the presence or absence of iron oxide at both sites A and B (Figures 2 & 3, $P < 0.001$). At 90 DAT, there was a significant drop in pH for all iron oxide affected plots: from 6.63 at 60 DAT, the pH dropped significantly to 4.67 for site A ($P < 0.001$), and from 6.22 at 60 DAT to 4.68 at 90 DAT for the site B ($P < 0.001$).

3.2. Assessment of the Intensity of Iron Toxicity (IRRI Scale)

Rice plants in plots not affected by iron oxide grew more

and were greener (Figure 4) compared to plants grown on plots affected by iron toxicity, yellowing of leaves (Figure 5).



Figure 4. Plants grown on plots not affected by iron toxicity at 60 days after transplanting.



Figure 5. Plants grown on plots affected by iron toxicity at 60 days after transplanting.

In plots not affected by iron oxide, for both sites, rice plants did not have iron toxicity symptoms at 0, 30 and 60 DAT (Figure 4, Table 2). It was only at 90 DAT, where, despite normal growth and tillering, brown spots appeared at the tip of the oldest leaves. In sites with the presence of iron oxide, there was no symptom of iron toxicity at 0 DAT. However, at 30 DAT, the older leaves turned orange-yellow. These symptoms intensified at 60 and 90 DAT, growth and tillering were reduced with many leaves becoming discolored for both sites (Figure 5, Table 2).

Table 2. Intensity of Iron Toxicity on rice plants at 0, 30, 60 and 90 days after transplanting (DAT) based on the IRRI Scale on sites A & B.

	Iron toxicity Intensity (Echelle IRRI)							
	Site A				Site B			
	0 DAT	30 DAT	60 DAT	90 DAT	0 DAT	30 DAT	60 DAT	90 DAT
Fields not affected by iron oxide								
P1	0	0	0	1	0	0	0	1
P2	0	0	0	1.3	0	0	0	0.6
P3	0	0	0	1	0	0	0	1
Fields affected by iron oxide								
PfeOx1	0	2.7	5.3	6.5	0	2.3	4.7	6.3
PfeOx2	0	3.7	5.7	6.3	0	3	4.3	6
PfeOx3	0	3.3	5.3	6.7	0	2.7	5	6.6

3.3. Effect of Iron Toxicity on Rice Yield

Rice yield was statistically higher in plots without iron oxide compared to plots with iron oxide. The yield was 2.21 t.ha⁻¹ in plots not affected by iron oxide compared to 1.28 t.ha⁻¹ in iron oxide affected plots (Figure 6). The presence of iron oxide led to a 43% decrease in rice yield in site A and a 49% decrease in site B.

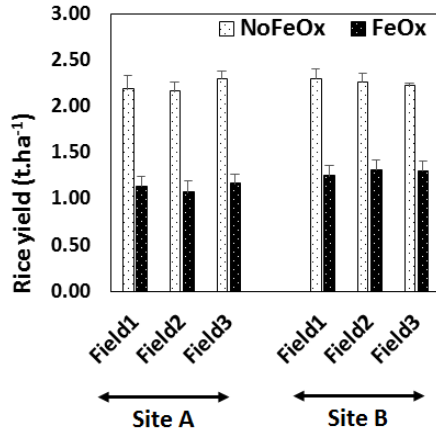


Figure 6. Rice yield on plots not-affected (NoFeOx) and affected (FeOx) by iron toxicity in both sites A and B.

4. Discussion

The acidic pH in all plots with or without iron oxide was characteristic of tropical ferruginous soils with leaching original materials [26]. However, at 0 and 90 DAT, the lower pH in the lowland paddy fields affected by iron toxicity compared to the pH in highland rice fields not affected by iron toxicity indicates that there was an acidification that affected the land differently. This is similar to results by [7, 27] who found that fields with iron oxide presented much lower pH compared to field not affected by iron toxicity. Indeed, in lowlands, water stagnates well beyond the rainy season, sometimes during the whole year. This stagnation of water creates anaerobic conditions where organic matter barely decomposes, increasing the acidity of the environment [28]. The difference in acidity at 0 DAT could also be due to a river deposit of sulfated material in the lowlands. Indeed, [4, 7, 26] found that in the some areas, ferric iron was inherent to the environment, and when there is water, ferric iron is reduced to ferrous iron contributing to acidifying the field. In anaerobic conditions, these sulfato-ferric compounds may be also reduced by sulfate-reducing bacteria which makes the plots more acidic [7]. When the pH drops to a certain level, below 5, Fe^{3+} ferric iron is reduced to ferrous iron Fe^{2+} which will be available to plants. Iron being a micronutrient becomes toxic even at very low doses when the threshold of absorption is exceeded [29]. Acidity could be implicated in the causes of iron toxicity. But we cannot isolate other causes of iron toxicity such as a nutritional imbalance in phosphorus, zinc because we have not measured the content of these elements.

At 30 and 60 DAT, the increase of pH could be due to a dilution of stagnant rainwater. This is in opposite to studies by [26] who found that the increase in rainfall resulted in submerging the paddy field and contributed to lowering even more the pH. Although the difference is not significant between 30 and 60 DAT, the slight downward trend in pH at 60 DAT could be due to much lower rainfall quantities and more intense evaporation, creating less diluted conditions. This decrease in rainwater may have also contributed to keeping the pH at a lower level at 90 DAT that may have

intensified the symptoms of iron toxicity and may have affected the development and yield of rice. Similar studies by [30, 31] have shown also that iron toxicity affected the growth of rice respectively trough root and shoot development. In conditions of weak iron toxicity, rice can sometimes develop resilience mechanisms but when the toxicity becomes intense, plants will not survive. Mechanisms of rice resilience against iron toxicity have been shown with studies by [32, 33, 34].

The presence of iron oxide hindered rice yield with a decrease of at least 43% in plots affected by iron toxicity. Lowland paddy fields affected by iron toxicity have been shown to have lower rice growth and yield [35]. This is in accord with studies by [26] who showed that plots with iron toxicity had lower rice yield compared to plots not affected by toxicity. In lowland, because of the presence of water, the iron is reduced and acidify further the soil. More iron will be available, plant nutrition will be affected and yield will decrease [36, 37]. Similar results were found by [38] who showed that depending on the form of iron and the severity of the toxicity, plant growth will be affected. Furthermore, [39], showed also that rice yield was reduced due to iron toxicity in acidic conditions. When the soil is highly acid, plant absorption of phosphorus and magnesium is reduced, this will affect plant growth [40]. Iron toxicity will cause lower rice yield and depending on the severity of the symptoms, plants will not survive. Our results showed that plots with intense iron toxicity symptoms were associated with the lowest yield. This is in opposite with results by [34, 41] who showed a decrease in rice yield even in plots where plants did not have symptoms of iron toxicity. Yield reduction were observed in plant without significant leaf bronzing [19, 23]. Iron toxicity can sometimes be reduced by applying better fertilization, particularly by adding phosphorus and potassium [42].

5. Conclusion

Iron toxicity is a major problem for rice growers in South Senegal. In this study, with the decrease of rainwater, lowland fields became more acidic in plots with the presence of iron oxide compared to plots not-affected by iron oxide. Symptoms of iron toxicity were more intense on plants grown in plots affected by iron oxide compared to plants grown in plots not affected by iron oxide. Rice yield decreased by at least 43% due to iron toxicity in plots with iron oxide compared to plots without iron oxide. Plots with the lowest rice yield were the one located in lowland areas with more acidic conditions. Given these results, practices to raise pH in paddy fields must be made. Soil amendments may be used to raise pH; this will also provide plants with nutrients. In addition to adopting better soil management practices, the use of resilient rice varieties may be recommended.

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